

### Polyatomic Ions

Compound	Name	Charge
$\text{NH}_4^+$	Ammonium	+1
$\text{H}_3\text{O}^+$	Hydronium	+1
$\text{OH}^-$	Hydroxide	-1
$\text{CN}^-$	Cyanide	-1
$\text{MnO}_4^-$	Permanganate	-1
$\text{ClO}_4^-$	Perchlorate	-1
$\text{ClO}_3^-$	Chlorate	-1
$\text{ClO}_2^-$	Chlorite	-1
$\text{ClO}^-$	Hypochlorite	-1
$\text{NO}_3^-$	Nitrate	-1
$\text{NO}_2^-$	Nitrite	-1
$\text{HCO}_3^-$	Hydrogen carbonate (bicarbonate)	-1
$\text{C}_2\text{H}_3\text{O}_2^-$ or $\text{CH}_3\text{COO}^-$	Acetate	-1
$\text{SO}_4^{2-}$	Sulfate	-2
$\text{SO}_3^{2-}$	Sulfite	-2
$\text{CO}_3^{2-}$	Carbonate	-2
$\text{CrO}_4^{2-}$	Chromate	-2
$\text{Cr}_2\text{O}_7^{2-}$	Dichromate	-2
$\text{S}_2\text{O}_3^{2-}$	Thiosulfate	-2
$\text{PO}_4^{3-}$	Phosphate	-3